

Variation Method- Helium

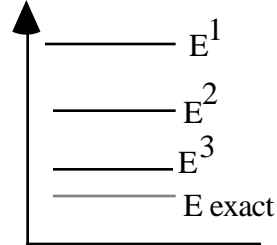
If you guess a solution to the Schrödinger Equation, how do you find out how good a guess you have made? $H\Psi = E \Psi$ can't be solved exactly:

but you have :

$\Psi^{(1)}$ guess

$\Psi^{(2)}$ better guess

$\Psi^{(3)}$ better yet



The Variation Theorem guarantees that the trial energy is always greater than the exact energy.

$$H\Psi_i = E_i \Psi_i$$

$$\int \Psi_i^* H \Psi_i d\tau = \int \Psi_i^* E_i \Psi_i d\tau$$

solve for E_i :
$$E_i = \frac{\int \Psi_i^* H \Psi_i d\tau}{\int \Psi_i^* \Psi_i d\tau} \quad \text{is exact}$$

$$E_i^{(1)} = \frac{\int \Psi_i^{(1)*} H \Psi_i^{(1)} d\tau}{\int \Psi_i^{(1)*} \Psi_i^{(1)} d\tau}$$

$E^{(1)} \geq E$ by the variation theorem

$$\Psi_{gs}^{(1)} = \Psi_1 \Psi_2 = \left(\frac{1}{\pi} \left(\frac{Z_{eff}}{a_0} \right)^3 \right) e^{-Z_{eff} r_1/a_0} e^{-Z_{eff} r_2/a_0}$$

$$E^{(1)} = \frac{\int \Psi_{gs}^{(1)*} H \Psi_{gs}^{(1)} d\tau}{\int \Psi_{gs}^{(1)*} \Psi_{gs}^{(1)} d\tau} = \frac{e^2}{a_0} \left(Z_{eff}^2 - \frac{27}{8} Z_{eff} \right)$$

minimize by changing Z_{eff} :

$$\frac{dE^{(1)}}{dZ_{eff}} = 0 = \frac{e^2}{a_0} (2 Z_{eff} - 27/8) \quad \text{see Karplus and Porter 4.1.5}$$

$$Z_{eff} = 27/16 = 1.6875$$

$$E = -77.5 \text{ eV} \quad (\text{exp. } -79 \text{ eV})$$